

# LIGHT Notes

We **see an object** in 2 ways:

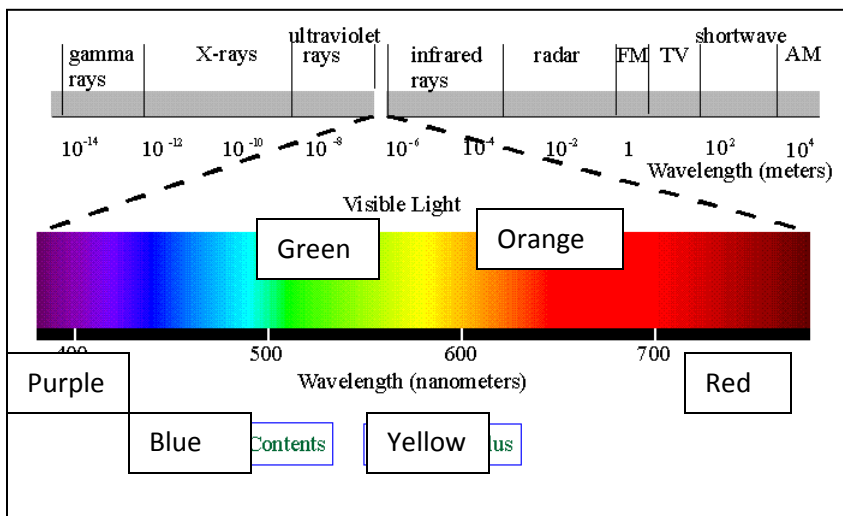
1. The object is the source of light

Ex:

2. By light reflected from an object which enters our eyes and allows us to see object

\_\_\_\_\_ : Energy in the form of electric and magnetic fields- called **electromagnetic radiation**. Acts like both a particle and a transverse wave as it travels.

Types of Electromagnetic Energy:



Electromagnetic Spectrum:

**TV, Radio:** long  $\lambda$ , low  $f$

**Infrared:** Heat

**Ultraviolet:** UV rays emitted  
Sun, tanning booths  
Birds, bees may see them

**X-rays:** Fires e- at Tungsten  
which releases x-rays

**Gamma rays:** high energy  
found in stars, cancer treat.

**Cosmic rays:** highest energy  
Solar flares, supernovas

**Light:**

Visible light is **electromagnetic radiation** with a  $\lambda$  of  $4 \times 10^{-7}$  m to  $7.5 \times 10^{-7}$  m

\*Travels in straight lines called \_\_\_\_\_

\*Does NOT need a \_\_\_\_\_ to travel through (why light travels in space)

\*Behaves like both a \_\_\_\_\_ and a \_\_\_\_\_

The speed of light is \_\_\_\_\_

Speed of light (c) = \_\_\_\_\_



If the wavelength of light increases, what will happen to the frequency of the wave?

Which color (purple or red) has a greater frequency? (more waves/sec) Explain your answer.

Name \_\_\_\_\_ Hour \_\_\_\_\_

**Spectroscope Lab:**

Use the spectroscope to find the wavelength of each colored line, and then use  $C = f \times \lambda$  to calculate the frequency of each. Determine what each element is from list below.

**Control/Practice: White Light**

**Unknown Element #1:** \_\_\_\_\_

**Unknown Element #2:** \_\_\_\_\_

**Unknown Element #3:** \_\_\_\_\_

The frequencies of possible elements:

**Mercury** (3 lines):  $6.7 \times 10^{14}$  Hz,  $5.5 \times 10^{14}$  Hz,  $5 \times 10^{14}$  Hz

**Hydrogen** (3 lines):  $7 \times 10^{14}$  Hz,  $6.1 \times 10^{14}$  Hz,  $4.3 \times 10^{14}$  Hz

**Neon** (3 thick lines, 1 faint line):  $5.5 \times 10^{14}$  Hz,  $5 \times 10^{14}$  Hz,  $4.6 \times 10^{14}$  Hz,  $4.5 \times 10^{14}$  Hz

### Take home lab: Calculate the speed of light

1. Completely cover the bottom of a microwave safe casserole dish with marshmallows.
2. Cook on low heat until you see some parts of the marshmallows start to melt.
3. Measure the distance between the melted spots: \_\_\_\_\_ **cm** = \_\_\_\_\_ **m**
4. The distance is equal to  $\frac{1}{2}$  a **wavelength**. What is the microwave  $\lambda$ ? \_\_\_\_\_ **m**
5. Determine the **frequency** of your microwave (use 2450 MHz if can't find it). \_\_\_\_\_ Hz
6. Use  $v = f \times \lambda$  to calculate the speed of light \_\_\_\_\_
7. Determine a % error for your calculation.  $\frac{(\text{acc-expt})}{\text{Acc}} \times 100\%$  \_\_\_\_\_

Must be done in a non-rotating microwave!